## Rowan University [Rowan Digital Works](https://rdw.rowan.edu/)

[Henry M. Rowan College of Engineering Faculty](https://rdw.rowan.edu/engineering_facpub)<br>Scholarship

Henry M. Rowan College of Engineering

1-12-2017

# Synthesis and analysis of separation networks for the recovery of intracellular chemicals generated from microbial-based conversions

Kirti Yenkie Rowan University

Wenzhao Wu University of Wisconsin-Madison

Christos Maravelias University of Wisconsin-Madison

Follow this and additional works at: [https://rdw.rowan.edu/engineering\\_facpub](https://rdw.rowan.edu/engineering_facpub?utm_source=rdw.rowan.edu%2Fengineering_facpub%2F49&utm_medium=PDF&utm_campaign=PDFCoverPages)   $\bullet$  Part of the [Biotechnology Commons](http://network.bepress.com/hgg/discipline/111?utm_source=rdw.rowan.edu%2Fengineering_facpub%2F49&utm_medium=PDF&utm_campaign=PDFCoverPages), and the [Chemical Engineering Commons](http://network.bepress.com/hgg/discipline/240?utm_source=rdw.rowan.edu%2Fengineering_facpub%2F49&utm_medium=PDF&utm_campaign=PDFCoverPages)

## Recommended Citation

Yenkie, K. M.; Wu, W.; Maravelias, C. T. (2017). Synthesis and analysis of separation networks for the recovery of intracellular chemicals generated from microbial-based conversions. Biotechnology for Biofuels 10(1), 119-140.

This Article is brought to you for free and open access by the Henry M. Rowan College of Engineering at Rowan Digital Works. It has been accepted for inclusion in Henry M. Rowan College of Engineering Faculty Scholarship by an authorized administrator of Rowan Digital Works.

## **RESEARCH**

**Open Access**



# Synthesis and analysis of separation networks for the recovery of intracellular chemicals generated from microbial-based conversions

Kirti M. Yenkie<sup>1</sup>, Wenzhao Wu<sup>1</sup> and Christos T. Maravelias<sup>1,2\*</sup>

## **Abstract**

**Background:** Bioseparations can contribute to more than 70% in the total production cost of a bio-based chemical, and if the desired chemical is localized intracellularly, there can be additional challenges associated with its recovery. Based on the properties of the desired chemical and other components in the stream, there can be multiple feasible options for product recovery. These options are composed of several alternative technologies, performing similar tasks. The suitability of a technology for a particular chemical depends on (1) its performance parameters, such as separation efficiency; (2) cost or amount of added separating agent; (3) properties of the bioreactor effluent (e.g., biomass titer, product content); and (4) fnal product specifcations. Our goal is to frst synthesize alternative separation options and then analyze how technology selection afects the overall process economics. To achieve this, we propose an optimization-based framework that helps in identifying the critical technologies and parameters.

**Results:** We study the separation networks for two representative classes of chemicals based on their properties. The separation network is divided into three stages: cell and product isolation (stage I), product concentration (II), and product purifcation and refning (III). Each stage exploits diferences in specifc product properties for achieving the desired product quality. The cost contribution analysis for the two cases (intracellular insoluble and intracellular soluble) reveals that stage I is the key cost contributor (>70% of the overall cost). Further analysis suggests that changes in input conditions and technology performance parameters lead to new designs primarily in stage I.

**Conclusions:** The proposed framework provides signifcant insights for technology selection and assists in making informed decisions regarding technologies that should be used in combination for a given set of stream/product properties and fnal output specifcations. Additionally, the parametric sensitivity provides an opportunity to make crucial design and selection decisions in a comprehensive and rational manner. This will prove valuable in the selection of chemicals to be produced using bioconversions (bioproducts) as well as in creating better bioseparation fow sheets for detailed economic assessment and process implementation on the commercial scale.

**Keywords:** Downstream separation, Physical property, Technology selection, Optimization, Cost contribution, Threshold value

## **Background**

Concerns regarding climate change, energy security, and petroleum costs have encouraged the search for

\*Correspondence: maravelias@wisc.edu

<sup>1</sup> Department of Chemical and Biological Engineering, University

Full list of author information is available at the end of the article

alternative and sustainable sources of energy, fuels, and chemicals  $[1-3]$  $[1-3]$  $[1-3]$ . Significant work has been done in the bioenergy and biofuel sectors [\[4](#page-19-2), [5\]](#page-19-3); however, strategies for the production of bio-based chemicals are at their infancy [\[6,](#page-19-4) [7](#page-19-5)]. Bio-based chemicals production has signifcant advantages, such as carbon neutrality and bioremediation, over traditional petrochemical routes [\[8](#page-19-6)[–10](#page-19-7)]. Furthermore, the feedstocks involved in petrochemical



© The Author(s) 2017. This article is distributed under the terms of the Creative Commons Attribution 4.0 International License [\(http://creativecommons.org/licenses/by/4.0/\)](http://creativecommons.org/licenses/by/4.0/), which permits unrestricted use, distribution, and reproduction in any medium, provided you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The Creative Commons Public Domain Dedication waiver ([http://creativecommons.org/](http://creativecommons.org/publicdomain/zero/1.0/) [publicdomain/zero/1.0/](http://creativecommons.org/publicdomain/zero/1.0/)) applies to the data made available in this article, unless otherwise stated.

of Wisconsin-Madison, 1415 Engineering Drive, Madison, WI 53706-1691, USA

processes are crude oil and natural gas. Natural gas (low cost) can contribute to the production of low-carbon content molecules such as methane, ethane, and propane. However, the production of higher carbon-containing molecules, such as butenes from natural gas, requires catalytic oligomerization which is non-trivial and cost intensive  $[10-12]$  $[10-12]$  $[10-12]$ . Typically, these high-carbon ( $\geq 4$ ) molecules are produced from cracking of naphtha, gas oils, or from crude refnery streams [[13,](#page-19-9) [14\]](#page-19-10). Since these feedstocks are quite expensive and have limited reserves when compared to natural gas, there is scope for substitution with bio-renewable feedstocks.

There are several bio-renewable sources such as biodegradable wastes, dedicated energy crops, lignocellulosic biomass, and microbial cultivations [\[15](#page-19-11)]. Microbial hosts are advantageous as they can be engineered precisely to produce molecules of interest and the process can be controlled using diferent bioreactor and fermenter configurations  $[16]$  $[16]$  $[16]$ . Also, their ability to use external carbon sources other than atmospheric carbon dioxide is an added advantage [\[17](#page-19-13)–[20\]](#page-20-0). Microbes can utilize energy in the form of light, inorganic and organic substrates for cellular growth, growth-independent cellular maintenance, and extracellular (secreted) product formation [[21\]](#page-20-1), as shown in Fig. [1.](#page-2-0) Cellular growth and extracellular product formation fuxes can be manipulated by metabolic engineering and/or controlled conditions [[22](#page-20-2)[–29](#page-20-3)]. The energy required for cellular maintenance cannot be manipulated since it is an essential function for cell survival.

Intracellular chemicals produced in microbial cultivations are usually energy or carbon storage compounds [[30\]](#page-20-4). They have diverse functions and properties which render them suitable for several industrial and specialty applications; examples include biopolymers [[31\]](#page-20-5) and biolubricants  $[32]$ . The knowledge about molecular reactions and cellular level synthesis pathways has made it possible to alter the concentration, properties, and structures of intracellular constituents to products of interest [[33](#page-20-7)[–38](#page-20-8)]. Continuous developments and improvements in the feld of metabolic engineering have generated hope that the commercialization of bio-based chemicals can be made economically feasible and proftable. Metabolic engineering tools (Fig. [1\)](#page-2-0) include pathway and enzyme engineering to create synthetic pathways, substrate utilization engineering, transporter engineering, byproduct elimination and precursor enrichment, and rerouting pathways [[23,](#page-20-9) [39](#page-20-10)[–41](#page-20-11)]. Some examples of industrial production of biopolymers after the application of the formerly mentioned techniques are Biopol (P(3HB-*co*-3HV)) produced by ICI, Zeneca and Monsanto [\[42](#page-20-12)], Nodax (short- and medium-chain length PHAs) by P&G and MHG [[43\]](#page-20-13), and P(3HB-*co*-3HHX) by Kaneka Corporation [\[44](#page-20-14)].

The selection of microbial cultivation method and increase in the product content in the cell are two important aspects of bioprocess development. Equally important, if not more, is the recovery of bioproducts in desired form and purity economically. Usually, product content is very low in the (bio)reactor effluent streams (less than 20 wt%) [\[45](#page-20-15)], and hence separation cost can amount to 60–80% of the overall production cost  $[46-49]$  $[46-49]$  $[46-49]$ . Also, genetic modifcations and growth conditions can result in the formation of complex streams containing constituents which can introduce additional challenges in

<span id="page-2-0"></span>

product recovery. For example, selection of a suboptimal microbial strain enables faster initial progress in terms of product titer, rate, and yield (TRY) enhancement, but results in the formation of co-products which might be difficult to isolate later  $[6]$  $[6]$ . Hence, designing of the overall process in terms of upstream considerations, resulting effluent (also addressed as process stream) stream as well as the commercial-scale separation process, should be considered from the very beginning while developing new processes [[50](#page-20-18), [51\]](#page-20-19).

Previous work on economic assessment for recovery of intracellular metabolites has been restricted to specifc examples (details in Additional fle [1](#page-19-14)) on storage polymers such as PHAs [[52](#page-20-20), [53\]](#page-20-21), cyanophycin [\[54](#page-20-22), [55\]](#page-20-23), and pigments such as astaxanthin [\[56](#page-20-24)[–58\]](#page-20-25) and β-phycoerythrin [[59\]](#page-20-26). Furthermore, assessment studies have been performed for individual separation technologies [[60–](#page-20-27)[63](#page-20-28)] and some guidelines have been suggested regarding their applicability to microalgae harvesting [[64\]](#page-20-29) and biofuel production [\[45](#page-20-15)]. However, a systematic framework for quantitative analysis and selection of technology alternatives is not available. Traditional analyses have usually focused on sensitivity studies where the technologies in the separation network are already fxed and one parameter is varied at a time to analyze its efect on the process economics [[65–](#page-20-30)[67](#page-21-0)]. However, technology selection is not trivial since many competing alternatives are often available. Hence, researchers have suggested alternative methods such as generation of schemes and superstructures for the synthesis of separation networks [[48,](#page-20-31) [68–](#page-21-1)[76\]](#page-21-2). A separation scheme [\[77](#page-21-3)] incorporates a list of technologies for product isolation from a mixture of components. A separation superstructure [[74,](#page-21-4) [78\]](#page-21-5) is a network-based representation of all potentially useful technologies and the interconnections among them, and it is used as a basis for the formulation of optimization models.

A separation scheme  $[77]$  $[77]$  starts with the effluent of a (bio)reactor and enables the generation of a separation superstructure, which can be used for systematic process synthesis of separation processes [[74,](#page-21-4) [78–](#page-21-5)[81](#page-21-6)]. To come up with systematic guidelines for designing separation networks in the future, the key objectives of this work are to:

- •• develop a methodology for the assessment of separation technologies performing similar tasks and analyze the efect of technology selection on overall process economics;
- •• include the complete separation network while performing the economic assessment so as to understand the combination of technologies in a selected network and the cost-intensive tasks;
- analyze the change in process cost with change in important process parameters such as biomass titer, cellular product content, product purity, and technology performance, based on their relative impact in the separation network selection; and
- •• determine the critical values of the selected parameters and understand the shifts in technology selection for designing efficient separation processes in the future.

In the ["Methods"](#page-3-0) section, we present the proposed analysis framework which includes formulation of separation networks, modeling, and solution strategy. In the "[Results and discussion"](#page-5-0) section, we present two case studies for intracellular products; the frst study represents intracellular insoluble products, while the second study represents intracellular soluble products. We propose a base case for each and then discuss technology selection and important process considerations. This is followed by further analysis and results for selected parameters in each product class. Toward the end, we present some key conclusions drawn from this work.

#### <span id="page-3-0"></span>**Methods**

In this section, we discuss the stage-wise separation scheme and specifc classes of intracellular products; the superstructure generation and solution method for these product classes; and the analysis framework for the assessment of separation technologies.

#### **Stage‑wise separation scheme**

The recovery of an intracellular bioproduct is divided into three stages: (I) Cell and product isolation, (II) Product concentration, and (III) Product purifcation and refnement. Each stage can have multiple technologies for performing similar tasks. Sometimes, more than one technology needs to be selected in a stage and sometimes a stage can be skipped, based on product properties, initial stream concentration, and desired recovery and purity constraints. In Fig. [2,](#page-4-0) we present a schematic of the three-stage separation framework for intracellular products. The different technologies applicable for each task are listed in Table [1](#page-4-1) [\[72](#page-21-7), [79,](#page-21-8) [82–](#page-21-9)[95\]](#page-21-10).

#### **Intracellular product classes**

The technology options available for various tasks listed in the three-stage separation scheme can be narrowed down depending on other distinguishing product properties such as the product's solubility in water [insoluble (NSL) or soluble (SOL)], physical state [solid (SLD) or liquid (LQD)], density with respect to water [heavy (HV) or light (LT)], relative volatility with respect to water [volatile (VOL) and non-volatile (NVL)] for soluble products,



## <span id="page-4-1"></span><span id="page-4-0"></span>**Table 1 Technology options available for performing the tasks listed in three separation stages**



*ATPE* aqueous two-phase extraction

and intended use [commodity (CMD) or specialty (SPC)]. Thus, intracellular chemicals can be categorized into specifc product classes based on their properties. Such classifcation helps in identifying the relevant tasks and technology options in each stage of the separation scheme.

#### **Superstructure generation and solution method**

The potential separation stages and the relevant technology options can be reduced using additional product

properties (discussed earlier in intracellular product classes). Hence, building upon the previous work on separation schemes [\[96](#page-21-11)] and superstructure-based synthesis of separation networks [\[97](#page-21-12)], we generate an appropriate separation superstructure for each class of product. The next steps are formulation of a superstructure optimization model, solution to identify the optimal separation network design, and economic assessment.

The optimization model is formulated as a mixed-integer non-linear programming (MINLP) problem, with binary variables denoting the active (1) and inactive (0) states of technologies present in the separation superstructure. The objective is to minimize the 'overall process cost,' which comprises feed, annualized capital, materials, consumables, labor, utility, and other costs (e.g., supervisory and overhead cost)  $[98]$  $[98]$ . The optimization model is formulated in GAMS 24.4.6 environment and solved using BARON [\[99\]](#page-21-14), a global optimization solver.

#### **Analysis framework**

After generating a separation superstructure for a product class, we have multiple technologies which can perform the same task and each technology has a performance metric which indicates its suitability over other parallel technologies. The MINLP optimization model comprises separation technology models, stream flows, and product recovery and purity constraints. We formulate a base case using reference values for parameters such as input conditions, technology efficiencies, material cost, and requirements, and solve the optimization problem to identify the key cost drivers. However, since these reference parameters afect the process economics, we perform additional analysis to study how variations in the values of the aforementioned parameters impact the overall process cost and technology selection.

The steps involved in the proposed analysis framework are the following:

*Step#1* Formulate a base case and solve it to determine the optimal separation network and the key cost contributors. Also, determine alternate (next best) confgurations, to decide which technologies are essential and which can be changed in the optimal design with little compromise in the process cost.

*Step#2* Vary a combination of parameters for the key cost contributing technologies (i.e., solve multiple optimization problems with varying parameter values) to determine the critical values when there is a shift in technology selection.

*Step#3* Extend the analysis to other product classes based on (1) the results for the representative case, if the same technologies are available for the other classes, or (2) the literature, individual technology considerations, and simulation tools [\[100\]](#page-21-15), if new technologies should be considered.

## <span id="page-5-0"></span>**Results and discussion**

#### **Study 1: intracellular insoluble products**

Intracellular (IN) insoluble (NSL) products can be further classifed as solid (SLD)/liquid (LQD), heavy (HV)/ light (LT), and commodity (CMD)/specialty (SPC). Based on these, we list the product classes and the correspond-ing separation options in Table [2.](#page-5-1) The difference between commodity and specialty products is their required purity grade and hence similar technologies are applicable to both, but the latter may require more than one purifcation and/or refning step to achieve the desired purity.

We choose intracellular (IN), insoluble (NSL), solid (SLD), heavy (HV), commodity (CMD) product as a representative class for most intracellular insoluble products. The proposed analysis framework is applied to the aforementioned product class.

#### **Separation superstructure for IN NSL SLD HV CMD product**

We start with the general separation scheme for an intracellular insoluble product (refer Additional fle [1\)](#page-19-14) and simplify it to suit the specifc class of IN NSL SLD HV CMD product. This simplified scheme is used to generate a separation superstructure as illustrated in Fig. [3](#page-6-0). Since the product is intracellular, we have four tasks in stage I: (1) pretreatment, (2) cell harvesting, (3) cell disruption, and (4) phase isolation. Pretreatment is optional and can be used to increase the efective size of the cells through focculation. Cell harvesting is used to separate the cells from water present in the bioreactor effluent stream. Cell disruption releases the desired product along with other non-product cellular materials (NPCM) such as proteins, carbohydrates, lipids, nucleic acids, and cell debris in the resultant stream  $[101, 106]$  $[101, 106]$  $[101, 106]$ . Thus, the phase isolation task for this class may require multiple steps. One of the steps is product-rich phase formation that can be accomplished by two alternatives: diferential digestion and solubilization.

Diferential digestion uses an agent for NPCM digestion leaving the product unafected in the process [[52](#page-20-20), [107](#page-21-18)], while solubilization [[108,](#page-21-19) [109](#page-21-20)] uses a solvent which can selectively dissolve the product, leaving NPCM as is in the stream. Isolation of product using diferential digestion already achieves substantial product concentration, and thus stage II is not required. However, if the product has been isolated using solubilization, then stage II is required to recover the product by precipitation using an anti-solvent. This is followed by membrane separation (microfltration) to separate the precipitated product from the liquid phase. The product obtained after the two separation stages may still contain small amounts of water, acids, solvents, and anti-solvent. Drying in stage III can remove these traces and achieve the desired dry solid state with required purity specifcations for the product.

<span id="page-5-1"></span>



*PHAs* polyhydroxyalkanoates

*Mbr-all* is inclusive of all types of membrane technologies (MF, UF, RO, NF) Full forms of the technology abbreviations used are listed earlier in Table [1](#page-4-1)



<span id="page-6-0"></span>(Flc), sedimentation (Sdm), centrifugation (Cnt,1,2,3,4), fltration (Ftt,1,2), acid hydrolysis (Ahy), enzyme lysis (Ely), bead mill (Bml), membrane processes [microfltration (MF,1,2,3,4) and reverse osmosis (RO,1,2)], diferential digestion (Ddg), solubilization (Slb), precipitation (Prc), and drying (Dry). An option for bypassing (Byp) a set of parallel technologies is included in stage I

Some important input parameters essential to comprehend the cost contribution and further analysis are provided in Table [3](#page-6-1).

#### **Cost contribution analysis**

The objective is to 'minimize the overall process cost.' Details regarding the input conditions and technology parameters for the base case and the MINLP problem formulation are discussed in Additional fle [1](#page-19-14).

#### *Optimal confguration*

The base case optimal configuration and cost contribu-tions are presented in Fig. [4.](#page-6-2) The technologies selected in stage I include focculation (Flc) for pretreatment,

#### <span id="page-6-1"></span>**Table 3 Important input parameters for the base case for IN NSL SLD HV CMD product**





<span id="page-6-2"></span>centrifugation (Cnt,1) for cell harvesting, acid hydrolysis (Ahy) for cell disruption, centrifugation (Cnt,2) for initial phase separation, and diferential digestion (Ddg) for product-rich phase formation, followed by centrifugation (Cnt,3) to separate the solid product from the digested NPCM and other liquid phase components. Stage II is bypassed because of the relative product concentration already achieved in stage I. The final product refinement in stage III is achieved by drying (Dry). The cost contribution shown in Fig. [4](#page-6-2) reveals that stage I is the key cost driver (73%). The relative contribution by different tasks in the separation process is also presented. The overall process cost is \$10.34/kg product where the separation cost contribution is  $$8.69/kg$  (~84%). The separation cost is a summation of the annualized capital, materials, consumables, utilities, labor, and other costs (refer Additional fle [1](#page-19-14)).

#### *Alternate confgurations*

To determine the alternate (next best) separation confgurations, we add successive integer cuts [\[110\]](#page-21-25) as additional constraints in our model (see Additional fle [1\)](#page-19-14). In Table [4,](#page-7-0) we present the next three alternate configurations along with the overall process cost and separation cost.

#### **Important process considerations**

Based on the base case cost contribution analysis, some key process parameters are identifed. Changes in the values of these parameters have the potential of afecting the optimal separation network design as well as process economics. The details of the selected parameters and their probable range of variation are presented in Table [5](#page-8-0).

*Biomass titer* in the feed entering the separation network is a parameter dependent on the microbial strain, cultivation route, substrate utilization, and bioreactor design. It has a potential to be altered by upstream features such as metabolic engineering tools [[23](#page-20-9), [111\]](#page-21-26). For example, microbial strains can be engineered to enhance the accumulation of desired product inside the cells [[112–](#page-21-27)[114](#page-21-28)]. Additionally, suitable microbial hosts that can tolerate stress conditions such as product or co-product toxicity, and growth inhibitors can achieve high product yields and biomass titers  $[115]$  $[115]$ . Thus, biomass titer can difer for diferent microbial strains and product systems, and hence it is selected for further analysis.

Cell harvesting technologies The performance for sedimentation and centrifugation is defined in terms of 'efficiency' of the separation of cells from the aqueous phase. For fltration, it is defned as the 'retention factor' of cells on the retentate side of the flter. It is dependent on cell size, relative cell concentration in the process stream, and properties such as hydrophobicity, relative density, and shear susceptibility [[85,](#page-21-30) [86,](#page-21-31) [116\]](#page-21-32). For example, a larger sized microbial cell may be separated effectively by 'filtration,' while a denser cell may be separated by simple 'gravity sedimentation.' For non-Newtonian flows (cell concentration >15%), the performance of sedimentation and centrifugation will be afected negatively as they both are dependent on viscosity, but fltration and membranes might perform better. The efficiency of sedimentation is dependent on Stokes' law [\[117\]](#page-21-33), proportional to gravitational acceleration and the square of particle diameter, and inversely proportional to viscosity. These are properties of the components in the system, and hence there is limited scope for performance enhancement when compared to centrifugation and filtration options  $[64]$  $[64]$ . Thus, efficiency of sedimentation  $(Sdm,1)$  is kept fixed at 70% (base case) and the values for centrifugation (Cnt,1) and fltration (Ftt,1) options are varied in the selected range (Table [5\)](#page-8-0) for analysis.

*Cell disruption technologies* After cell harvesting, the subsequent step is cell disruption. The concentration of cells in the stream entering the cell disruption technology has a considerable efect on the overall cost as well as product release. Bead mill is a mechanical method for cell disruption and does not introduce any secondary agents in the system. However, the properties like cell wall thickness, cell size, and shear susceptibility can change the product release efficiency [\[64](#page-20-29), [124](#page-22-0), [125](#page-22-1)]. Chemical and enzymatic lyses are other cell disruption methods which have gained popularity due to higher product release efficiency, selectivity, and low energy requirements. However, these methods introduce other components in the system, which increases the amount of materials handled downstream. The potential to recover and recycle the enzymes and chemicals in these methods is an important issue which needs further research. Enzymatic lysis can result in smaller cell debris particles which can be difficult to isolate later. All these parameters and considerations can alter the product release efficiency of the cell disruption technologies. Hence, in the third analysis, we choose the performance of cell disruption technologies, defned in terms of the percentage release of intracellular

<span id="page-7-0"></span>**Table 4 Alternate separation confgurations for study 1 using successive integer cuts**

Overall cost (\$/kg) (% increase) Separation cost (\$/kg)
8.69
9.05
9.41
9.36

<span id="page-8-0"></span>**Table 5 Parameters selected for analysis in study 1**

Separation stage and task	Parameter	Base case (nominal)	Range	<b>References</b>
Stage I: input parameter	Biomass titer in input process stream 5 $q/L$		$(0.1 - 200)$	$[45, 118 - 120]$
Stage I: task-cell harvesting	Cell separation efficiency for centrifugation (Cnt,1)	80%	$(70 - 95)$	[64, 121]
	Cell retention factor for filtration (Ftt,1)	80%	$(70 - 95)$	[85]
Stage I: task-cell disruption	% Release for acid hydrolysis (Ahy)	85%	$(80 - 98)$	$[122]$
	% Release for bead mill (Bml)	85%	$(80 - 98)$	[123]
	% Release for enzyme lysis (Ely)	90%	$(85 - 98)$	$[124]$
Stage I: task-phase isolation	Amount of agent required for differential digestion (Ddg)	0.4 kg/kg other cell components	$(0.05-3)$	[109]
	Amount of solvent required for solubilization (SIb)	0.4 kg/kg product	$(0.05-3)$	$[107]$

components for the bead mill (Bml), acid hydrolysis (Ahy), and enzyme lysis (Ely).

*Phase isolation technologies* The performance of differential digestion and solubilization is infuenced by the amounts and costs of digestion agent or solubilizing solvent added  $[52, 107, 108]$  $[52, 107, 108]$  $[52, 107, 108]$  $[52, 107, 108]$  $[52, 107, 108]$ . Thus, we select the materials added for these parallel technologies for further analysis. Stage II is absent for the base case and stage III contributes 11% in the overall cost. Consequently, we do not perform additional analysis for stage II and stage III parameters for this product class.

#### **Analysis and results for IN NSL SLD HV CMD product**

The results from the proposed framework are presented and some insights regarding shifts in technology selection and changes in optimal separation network design are provided.

#### *Biomass titer*

We vary the biomass titer in the range of 0.5–200 g/L as it includes the range for intracellular insoluble products in photoautotrophic  $(0.5-10 g/L)$  [\[118](#page-21-34), [120](#page-22-2)] as well as heterotrophic  $(10-200 \text{ g/L})$  [[53,](#page-20-21) [119\]](#page-21-35) conditions.

In Fig. [5,](#page-8-1) we show the variation in the overall process cost (axis Y1) and cost contributions (axis Y2) for feed, separation stages (Fig. [5a](#page-8-1)) and three important tasks of cell harvesting, cell disruption, and product isolation in stage I (Fig. [5b](#page-8-1)). We observe that the optimal separation network design changes with the change in titer values. When the titer is less than  $5 \text{ g/L}$ , design A is optimal; from 5 to less than 15 g/L, design B is optimal; and from 15 to 200 g/L, design C is optimal.

The selected technologies and active streams are highlighted in Fig. [6.](#page-9-0) For lower biomass titers (design A), enzyme lysis (Ely) proves to be the optimal choice



<span id="page-8-1"></span>contribution (Y2) by feed, and active separation stages I and III. **b** Cost contribution (Y2) by the three tasks in stage I: cell harvesting, cell disruption, and product isolation. The *vertical dotted lines* at 5 and 15 g/L represent the change in optimal separation network design



<span id="page-9-0"></span>as compared to acid hydrolysis (Ahy) in the base case (design B) because the enzyme required for disruption depends on the amount of biomass [[124](#page-22-0), [125](#page-22-1)], while acid is added to maintain a certain normality in the aqueous phase [\[122\]](#page-22-4). For titer values of 15 and greater, the microfltration (MF,2) option is selected after diferential digestion (Ddg) as compared to centrifugation (Cnt,3) in the base case (design B) because microfltration (MF,2) has a better retention and concentration factor as compared to centrifugation (Cnt,3). Along with the changes in optimal network design, we also observe changes in the dominant cost drivers (Fig. [5b](#page-8-1)). For titers less than 2.5 g/L, cell harvesting is the major cost contributor because of the high utility and capital costs of centrifugation (due to the large incoming flow). For titers between 2.5 and 30  $g/L$ , cell disruption is the major contributor as the amount of acid needed for cell disruption is high. For titers greater than 30 g/L, cell harvesting and disruption costs are low, so phase isolation becomes the major contributor as differential digestion (high digestion agent cost) is followed by microfltration (high consumable cost).

#### *Cell harvesting technologies*

The second analysis is performed by varying the cell separation efficiency for centrifugation and cell retention factor for filtration  $[64, 85, 121]$  $[64, 85, 121]$  $[64, 85, 121]$  $[64, 85, 121]$  $[64, 85, 121]$  $[64, 85, 121]$ . During this analysis, we assume that the sedimentation efficiency is constant at 70% since there is not much scope for efficiency improvement.

In Fig. [7](#page-10-0), the overall process cost per kilogram of product is shown as a function of the centrifuge efficiency and filtration retention factor. The black asterisk corresponds to the base case (centrifuge efficiency: 80% and filtration retention factor: 80%). The results show that centrifugation is the preferred technology for biomass harvesting in most cases (corresponding region shown by the vertical contour lines). Filtration gets selected when the retention factor is greater than 80% and the corresponding centrifuge efficiency is less than 76%. Sedimentation is selected when centrifuge efficiency is less than  $77.5%$ , and filtration retention factor is less than 82%.

Centrifugation is the preferred technology in most cases because its major cost contributor is utility, which is lower than the large capital cost of the sedimentation tank and the high consumable replacement cost of the fltration membrane. Also, in some cases, although individual centrifugation may be costly, its combination with other technologies renders a lower cost (due to increased centrifugation efficiency and decreased input flow rate into other technologies).

#### *Cell disruption technologies*

The analysis for cell disruption technologies is performed by varying the percentage release of intracellular

![](_page_10_Figure_1.jpeg)

<span id="page-10-0"></span>components for the three technology options: bead mill, acid hydrolysis, and enzyme lysis [\[122](#page-22-4)[–124\]](#page-22-0). All the three available technologies can have variable component releases depending upon the type of microbial biomass handled, the entering feed characteristics, and product sensitivity to harsh conditions. The overall process cost as a function of the percentage release in acid hydrolysis and enzyme lysis is shown in Fig.  $8$ . The base case is denoted by the black asterisk at coordinates (85, 90). In most cases, acid hydrolysis is the preferred option due to the low cost of acid in comparison with the high cost of enzymes.

Furthermore, we observe that the option of bead mill was not selected for this analysis as its cost (14.94\$/kg at 98% release to 17.11\$/kg at 80% release) was much higher when compared to other two disruption options.

#### *Phase isolation technologies*

The analysis for phase isolation technologies is performed by varying the amounts of agent required in differential digestion and solvent required in solubilization. The overall process cost as a function of the variation in amounts of digestion agent and solubilizing solvent is shown in Fig. [9.](#page-10-2) Solubilization selection adds additional cost in stage II because of the requirement of precipitation and microfiltration technologies. Thus, even if a digestion agent is required in higher amounts as compared to solvent, it is still preferable to select diferential digestion rather than solubilization in most cases.

![](_page_10_Figure_6.jpeg)

<span id="page-10-1"></span>**Fig. 8** Overall process cost with variation in performance of cell disruption technologies. Direction of the contour lines denotes the viable region for the two technologies available for cell disruption: acid hydrolysis and enzyme lysis. They are *horizontal* where enzyme lysis is selected, whereas *vertical* where acid hydrolysis is selected. Critical values, when there is a change in technology selection from acid hydrolysis to enzyme lysis, are shown by *white lines*

![](_page_10_Figure_8.jpeg)

#### <span id="page-10-2"></span>**Extension to other classes of IN NSL products**

As per separation heuristics [\[84,](#page-21-36) [96\]](#page-21-11), the most plentiful impurity for an intracellular product is the excess amount of water present in the form of extracellular liquid. Thus, cell harvesting is the foremost task for all classes of intracellular products. This task needs to be followed by cell disruption to release the product from the intracellular matrix. After disruption, the product needs segregation from other cell constituents. Generally, insoluble solids tend to separate out with the cell debris and hence require product segregation methods such as diferential digestion and solubilization. Sometimes, if the solids are less dense than water (light) and have a relatively large particle size, they might easily float to the surface. Thus, low-cost technology, such as decantation, might suffice the product phase isolation task. Selective solubilization would be more favorable for lighter solids with smaller particle size.

Insoluble liquids that are lighter than water can be separated from solid impurities or heavier liquids using sedimentation, decantation, or centrifugation, which are usually low-cost technologies when less amount of materials are handled. Insoluble liquids if heavier than water might segregate along with the heavy solids and cell debris. Such products can be isolated using fltration or membranes depending upon the size of solid impurities. Thus, in the other three classes of intracellular insoluble products (IN NSL SLD LT CMD/SPC, IN NSL LQD LT CMD/SPC, and IN NSL LQD HV CMD/SPC), major differences occur in product isolation. The technologies are either similar to the ones discussed in the selected case study (solubilization or diferential digestion) or simple low-cost options such as decantation, sedimentation, centrifugation, fltration, or membranes. Furthermore, the results for biomass titer suggest that the cost contribution of the phase isolation task is always less than 20% in the overall process cost. Thus, it is safe to say that the proposed framework and the results presented for the representative case study can help in deciding the optimal separation confgurations for most intracellular insoluble products.

The current study does not consider parameters in stage II and stage III for further analysis as they do not contribute signifcantly in the separation of most intracellular, insoluble, high-volume chemicals. We did not include the scenario for high-value chemicals, because for these chemicals quality is a major concern and cost minimization becomes secondary [\[126](#page-22-6)]. However, for high-value chemicals a similar analysis can be performed to determine the impact of variation in technology parameters in the later separation stages II and III.

#### **Study 2: intracellular soluble products**

We list the product classes for intracellular (IN) soluble (SOL) products (based on additional properties) and the separation options involved for each of them in Table [6](#page-11-0). From the list of technologies for intracellular insoluble (Table [2\)](#page-5-1) and intracellular soluble (Table [6\)](#page-11-0) products, it is evident that the major diferences are observed in stage II. Thus, we analyze the different options in stage II for understanding their efects in the overall separation process. For this study, we choose the class of intracellular (IN), soluble (SOL), liquid (LQD), volatile (VOL), and specialty (SPC) product.

**Separation superstructure for IN SOL LQD VOL SPC product** The superstructure (Fig. [10\)](#page-12-0) for IN SOL LQD VOL SPC product is developed from the general separation scheme [[96\]](#page-21-11) for all intracellular soluble products (see Additional fle [1](#page-19-14)). We have fve tasks in stage I: pretreatment-1, cell harvesting, cell disruption, pretreatment-2, and phase isolation. Pretreatment #1 and #2 are optional while the other three tasks are essential for all IN SOL products. We do not consider acid hydrolysis for cell disruption because previous literature [\[122,](#page-22-4) [131\]](#page-22-7) does not support its suitability for most soluble products.

Due to low biomass titers, the product is usually present in dilute concentrations in the stream leaving stage I. Thus, the product is first concentrated in stage II using methods like distillation, aqueous two-phase extraction (ATPE) that involves a polymer–salt system [\[132](#page-22-8), [133](#page-22-9)], or liquid–liquid extraction that involves an extraction solvent [[134](#page-22-10)]. Distillation can concentrate the product by utilizing thermal energy (heat), whereas ATPE and extraction use mass separating agents, which require further separation from the product as well as recycling of the added agents. Tus, ATPE is followed by a membrane technology of ultrafltration to recover product from polymer phase, and extraction is followed by distillation to recover the product from the solvent. For further

<span id="page-11-0"></span>**Table 6 Product classes for intracellular and soluble type of product**

<b>Product classes</b>		Product examples Stage I technologies	<b>Stage II technologies</b>	Stage III technologies	References
IN SOL SLD (CMD/SPC)	Soluble proteins	Flc, Sdm, Ftt, Cnt, Ely, Bml, Hph, Mbr-MF/RO/UF, Slb	Mbr-all, Evp, Cnt, Prc, Ads	Mbr-all, Blc, Crys, Dry	[127, 128]
IN SOL LQD VOL (CMD/SPC)	Some algal biofuels	Flc, Sdm, Ftt, Cnt, Ely, Bml, Mbr-MF/RO/UF. Slb	Mbr- all, Ext, Dst, Atpe, Evp, Ads. Pvp	Mbr-all, Chr, Blc, Pvp	$[129]$
IN SOL LQD NVL (CMD/SPC)	Soluble hormones	Flc, Sdm, Ftt, Cnt, Ely, Bml, Mbr-MF/RO/UF. Slb	Mbr- all, Ext, Dst, Atpe, Evp, Ads, Pvp	Mbr-all, Chr, Blc, Pvp	[130]

*Mbr-all* is inclusive of all types of membrane technologies (MF, UF, RO, NF)

Full forms of the technology abbreviations used are listed earlier in Table [1](#page-4-1)

![](_page_12_Figure_1.jpeg)

<span id="page-12-0"></span>liquid (LQD) volatile (VOL) specialty (SPC) product. The three stages are distinguished based on *colors*: (I) Cell and product isolation: *red*; (II) Product concentration: *green*; and (III) Product purifcation and refnement: *blue*. The technologies involved are focculation (Flc,1,2), sedimentation (Sdm,1,2), centrifugation (Cnt,1,2), filtration (Ftt,1,2), enzyme lysis (Ely), bead milling (Bml), membrane processes (microfltration (MF) and ultrafltration (UF)), distillation (Dst,1,2), aqueous two-phase extraction (Atpe), extraction (Ext), chromatography (Chr), pervaporation (Pvp), and bleaching (Blc). Options for bypassing (Byp,1,2) a stage or some tasks are also included

purifcation and refning of the product, technologies like chromatography, pervaporation, and bleaching are available in stage III. Some basic input parameters for the case study are presented in Table [7](#page-12-1).

#### **Cost contribution analysis**

The objective is the same as in study  $1$ , to minimize the overall process cost. The details regarding input conditions, technology parameters, and MINLP problem formulation are discussed in Additional fle [1](#page-19-14).

#### *Optimal confguration*

The base case optimal configuration and the cost contributions are presented in Fig.  $11$ . The technologies

<span id="page-12-1"></span>![](_page_12_Picture_460.jpeg)

![](_page_12_Picture_461.jpeg)

selected in stage I include focculation (Flc,1) for pretreatment, centrifugation (Cnt,1) for cell harvesting, enzyme lysis (Ely) for cell disruption, flocculation (Flc,2) for pretreatment-2, and centrifugation (Cnt,2) for phase isolation. In stage II, distillation (Dst,1) is selected for product concentration. Pigments can sometimes impart undesirable color and appearance to the product. Hence, in stage III, bleaching (Blc) is used if the product is required to be colorless. Stage I is the key cost driver  $(-78.4%)$ . The overall process cost is \$11.29/kg product, wherein the separation cost contribution is \$9.57/kg  $(*85\%).$ 

#### *Alternate separation confgurations*

Similarly to case study 1, we use integer cuts successively to determine alternate separation confgurations (see Table [8](#page-13-1)).

#### *Alternate concentration options in stage II*

Distillation (Dst1) has been selected in the base case because the product is assumed to be volatile (relative volatility: 2.5) as compared to water. However, if the relative volatility is less than 1.05 then distillation is not preferred  $[135]$ . Thus, we also analyze the separation confgurations when ATPE and extraction are selected in stage II.

*Option #2*: *Aqueous two-phase extraction* In this case, we choose the aqueous two-phase extraction (Atpe) option in stage II (refer Additional file  $1$ ). The technologies selected (Fig.  $12a$ ) in stage I include flocculation (Flc,1) for pretreatment, sedimentation (Sdm,1) for cell harvesting, enzyme lysis (Ely) for cell disruption, and focculation (Flc,2) for second pretreatment, followed by sedimentation (Sdm,2) for phase isolation. In stage III, the remaining impurities are removed by pervaporation (Pvp) and bleaching (Blc) to achieve the desired product purity. The overall process cost is \$15.36/kg product wherein the separation cost contribution is \$13.09/kg  $(-85%)$ . Stage I is still the major cost contributor, followed by feed, stage III, and stage II.

*Option #3*: *Extraction* In this case, we choose liquid–liquid extraction (Ext) as the concentration option

![](_page_13_Figure_2.jpeg)

<span id="page-13-0"></span>in stage II and the following distillation (Dst,2) option for solvent recovery and recycling. The technologies selected (Fig.  $12b$ ) in stage I include flocculation (Flc,1) for pretreatment, centrifugation (Cnt,1) for cell harvesting, enzyme lysis (Ely) for cell disruption, and flocculation (Flc,2) and centrifugation (Cnt,2) for phase isolation. In stage III, the remaining impurities are removed by pervaporation (Pvp) and bleaching (Blc) to achieve the desired product purity. The overall process cost is \$23.4/ kg product wherein the separation cost contribution is \$21.49/kg (~92%). Stage II becomes the key cost driver, followed by stage I, feed, and stage III.

#### **Important process considerations**

Based on the base case confguration and other options in stage II, some key process parameters are identifed. Details of the selected parameters and their range of variation are presented in Table [9](#page-14-1).

*Biomass titer and product content* We select the input parameters of biomass titer and product content [[112](#page-21-27), [115](#page-21-29)] for the first set of analysis. The biomass titer has been varied in the range of  $0.1-250$  g/L as it is the feasible range reported for intracellular soluble products for both photoautotrophic (0.1–10) [[59\]](#page-20-26) and heterotrophic (5–268) [\[136](#page-22-16)] routes of microbial cultivation. Also, product content in the biomass has been varied in the range of 10–70% CDW [\[59,](#page-20-26) [137](#page-22-17), [138\]](#page-22-18).

*Parameters in stage I* Important considerations for the tasks in stage I are similar to the ones discussed for case study #1, since the major distinguishing property is product localization. The cell harvesting options are the same as in case study #1, and hence we do not perform additional analysis. For cell disruption, we check the overall process cost when bead mill is selected and it is much higher (15.29\$/kg at 98% release to 18.59\$/kg at 80% release) as compared to enzyme lysis. Hence, bead mill is never selected when all the other parameters are at their nominal values (refer Table [7](#page-12-1) and Additional fle [1](#page-19-14)).

*Parameters in stage II* For soluble products, stage II can be crucial as it involves concentration of the product present in water-rich phase along with other soluble components. The properties like physical state and volatility play a major role in separating soluble products. Technologies using energy separating agent (ESA) like distillation can prove efective if the relative volatility of the product is high (>1.05—for the more volatile component) and the heat of vaporization is low. However, if these conditions are not satisfed, then technologies using mass separating agents (MSA) like ATPE (Atpe) and extraction (Ext) have to be used. The availability of a suitable MSA, ease of product purifcation, and MSA recycling are important considerations for the ATPE and extraction options in stage II.

<span id="page-13-1"></span>**Table 8 Alternate separation confgurations for study 2 using successive integer cuts**

Configuration #	<b>Technologies selected</b>	Overall cost (\$/kg)	Separation cost (\$/kg)	
Best case	Flc1, Cnt1, Ely, Flc2, Cnt2, Dst1, Blc	$11.29$ (NA)	9.57	
2nd best	Flc1, Cnt1, Ely, Flc2, Sdm2, Dst1, Blc	11.38 (0.79%)	9.66	
3rd best	Flc1, Cnt1, Ely, Flc2, Ftt2, Dst1, Blc	14.29 (26.57%)	12.14	
4th best	Flc1, Sdm1, Ely, Flc2, Sdm2, Atpe, UF, Pvp, Blc	15.36 (36.04%)	13.09	

![](_page_14_Figure_2.jpeg)

<span id="page-14-0"></span>UF are chosen in stage II. **b** Technologies when Ext and Dst,2 are chosen in stage II. The active streams are shown by *bold red lines* and the selected technologies are highlighted in *diferent colors* corresponding to each stage: *red* for stage I, *green* for stage II, and *blue* for stage III. Stage-wise cost distribution assessment is also shown

<span id="page-14-1"></span>![](_page_14_Picture_450.jpeg)

![](_page_14_Picture_451.jpeg)

We select parameters from these technologies and vary them in the range reported in Table [9](#page-14-1) (refer Additional fle [1\)](#page-19-14). For distillation (Dst,1), the utility cost is the highest contributor, and hence we select the heat of vaporization of the product for further analysis. For ATPE (Atpe), the annualized capital cost is the highest contributor. The equipment size is a function of the feed and added materials during the separation process. We choose the

partition coefficient for the top phase (KpT) as the varying parameter since it afects the amount of polymer (top phase) and salt (bottom phase) added in the ATPE unit. For extraction (Ext), the materials cost is the highest contributor, and hence we select the solvent-specifc parameters of product partition coefficient  $(Kp)$  in solvent versus water, solubility of solvent in water, and the cost of solvent per unit mass for further analysis.

*Parameters in stage III* For specialty products, technologies in stage III have to be more stringent to meet the purity requirements. Based on the technologies used in the previous stages, the product-containing stream can have diferent components (other soluble materials and added separating agents such as solvents, polymers, and salts) and concentrations and hence one or more technology may be required to achieve the desired purity. The intracellular pigments impart undesirable color and appearance to the product; hence, bleaching is an important technology for specialty products. If chromatography is used for product purifcation, then the resulting stream will be dilute. Hence, pervaporation can be useful for the removal of excess water or solvents and further refnement of the product. However, the stage III parameters are not selected for further analysis as the cost contribution is less than 3% in the base case.

## **Analysis and results for IN SOL LQD VOL SPC product** *Biomass titer and product content*

*Biomass titer* The results of the analysis are shown in Fig. [13](#page-15-0), wherein the overall process cost (axis *Y*1) as well as the cost contributions (axis *Y*2) of the feed and the three separation stages are presented. We observe that the optimal separation network design changes with the titer. When the titer is less than 5 g/L, design A is optimal; from 5 to less than 15 g/L, design B is optimal; and from 15 to 250  $g/L$ , design C is optimal. The selected technologies for each design and active streams are highlighted in Fig. [14.](#page-16-0)

![](_page_15_Figure_4.jpeg)

<span id="page-15-0"></span>tion stages with varying biomass titer. Axis *Y*1 represents the overall process cost per unit product. Axis *Y*2 represents the cost contribution of the feed and three separation stages. The *vertical dotted lines* at 5 and 15 g/L represent the change in optimal separation network design

For lower biomass titers, Design A with sedimentation options in stage I prove to be optimal when compared to centrifugation options in the base case (Design B). This is because the low biomass titer in the input stream requires more materials to be handled in stage I, which makes centrifugation and fltration more expensive than sedimentation, even though it offers a lower biomass recovery. For titer values of 15 and greater, the bead mill (design C) is selected for cell disruption because of the trade-of between annualized capital cost (high for Bml and low for Ely) and material cost (none for Bml and high for Ely). As titer increases, the entering feed reduces in both bead mill and enzyme lysis. However, the enzyme required in enzyme lysis for a fxed production level remains constant even if the feed amount reduces. Thus, the materials handled will always be more in enzyme lysis when compared to bead mill.

*Product content* The overall cost as a function of product content is shown in Fig. [15a](#page-16-1). We observe a decrease in the overall process cost with the increase in product content. This happens because the higher the product content, the lower the amount of entering stream in the separation network for maintaining a constant production level. This decreases the feed cost as well as the cost incurred in each separation stage. The bar chart in plot (B) shows the contribution of feed and separation stages in the overall cost at varying product content, and this remains relatively constant for feed, stage I, and stage II and increases slightly with the increasing product content for stage III. This is because the technologies selected and the active process streams are the same. The lower amount of feed entering the separation network decreases the amount of materials handled in stages I and II; however, the materials handled in stage III remain fairly constant which increases its percentage contribution slightly with the increasing product content.

*Simultaneous variation in biomass titer and product content* We vary biomass titer (1–100 g/L) and product content (10–70%CDW) as shown in Fig. [16.](#page-17-0) We observe that design A (refer Fig.  $14$ ) is optimal for very low biomass titer values irrespective of the product content. Design B is optimal for titer values up to 70 g/L but with low product content ( $\leq$ 20%CDW at 5 g/L). The optimal separation network changes to design C for most of the biomass titer and product content values.

#### *Heat of vaporization of product*

The overall process cost increases with the increase in heat of vaporization (Additional file [1](#page-19-14)). This increase is not signifcant since the percentage contribution of distillation in the overall process cost is just ~4%. However, the stage II cost can increase by almost 6% if the heat of vaporization increases to 2200 kJ/kg from 700 kJ/kg (see

![](_page_16_Figure_1.jpeg)

<span id="page-16-0"></span>![](_page_16_Figure_2.jpeg)

<span id="page-16-1"></span>Fig. [17\)](#page-17-1). This happens because utilities are the major cost drivers in stage II; the more the heat of vaporization of the volatile component (product), the more heating utility will be required to perform distillation.

## **Partition coefficient for ATPE**

In Fig. [18](#page-17-2), we observe that the overall process cost decreases with the increase in partition coefficient. The increasing partition coefficient enables more amount of product to be extracted in the top phase. Thus, less amount of separating agents, i.e., polymer and salt, are required for product concentration. Along with the decrease in overall process cost, we also observe a change in the optimal separation design at KpT values of 6 and 8, and thus instead of a smooth decrease we see the trend change at the aforementioned KpT values.

## *Extraction parameters*

The change in partition coefficient (Kp) does not have a signifcant efect on the overall process cost (refer Additional file  $1$ ). The product recovery is a fixed parameter for the extraction technology; hence, at low Kp the increased equipment size and additional number of extraction stages contribute to the desired product recovery. As Kp increases, the equipment size and number of extraction stages decrease, hence resulting in slightly lower cost at higher Kp values. However, this change is not signifcant when compared to materials cost contributed by the added solvent.

The result for the solubility of solvent in water and the cost of solvent per unit mass is shown in Fig. [19.](#page-17-3) The nominal case of high solubility (0.03 kg solvent/kg water) and high solvent cost (1.5 \$/kg) has an overall process

![](_page_17_Figure_1.jpeg)

<span id="page-17-0"></span>**Fig. 16** Overall process cost as a function of biomass titer and product content. The change in optimal separation network design is denoted by the *white lines* and the details of the selected technologies and active stream in designs *a*, *b*, and *c* are the same as those represented in Fig. [14](#page-16-0). The base case (5 g/L, 20% CDW) is denoted by the *black asterisk*

![](_page_17_Figure_3.jpeg)

<span id="page-17-1"></span>cost of 23.5  $\frac{1}{2}$  /kg. This can be reduced to 12.6  $\frac{1}{2}$ /kg if the solubility decreases to 0.0002 kg/kg and the solvent cost decreases to 0.2 \$/kg. Thus, finding a solvent with low water solubility and low cost even if the partition coefficient values are low is beneficial. This is because low solubility will enable higher recycling and less solvent will be required to replenish the solvent loss in rafnate phase.

#### **Extension to other classes of IN SOL products**

For intracellular soluble products, additional properties such as physical state and volatility are important while separating the product from other components. In the

![](_page_17_Figure_7.jpeg)

<span id="page-17-2"></span>![](_page_17_Figure_8.jpeg)

<span id="page-17-3"></span>case study, we considered a liquid product which is not temperature sensitive. Hence, the volatility gradient plays a major role in deciding the optimal product concentration technology. If the desired product is in solid state, then membranes, evaporation, precipitation, or adsorption can be used for product concentration. Membranes and precipitation costs (refer case study #1, intracellular insoluble chemicals) are lower when compared to evaporation (energy intensive) and adsorption (materials costs and regeneration of adsorbent).

The purification and refinement technologies may include crystallization and/or drying, based on market requirements. Drying cost can be in a similar range reported for case study #1 ( $\sim$ 11% contribution); however, crystallization cost is dependent on the product's solubility as a function of temperature (cooling or evaporative crystallization), heat of crystallization, and structure-specifc properties such as chiral composition [\[145\]](#page-22-23) and preferred crystal shape. Usually, evaporative crystallization is expensive as compared to cooling crystallization, since it involves two operations, evaporation followed by crystallization. Cooling crystallization requires a substantial variation in solubility with temperature and its cost contribution can range from  $\sim$ 9% [[80](#page-21-37)] to 26% (Additional file [1](#page-19-14)) based on the properties of the product.

#### **Conclusion**

We studied separation systems for the recovery of intracellular chemicals produced by microbial conversions and presented two representative case studies that include most technologies which are common to other intracellular products.

The first study (IN NSL SLD HV CMD) is a representative for intracellular, insoluble chemicals. In this study, we found that product separation contributed to almost  $\sim$ 84% in the overall process cost and stage I was the major cost driver with ~73% cost contribution. Further analysis suggests that the overall cost can decrease signifcantly with an increase in biomass titer up to 50 g/L, a point beyond which the overall cost remains fairly constant. Major shifts in technology selection occur for the tasks of cell disruption and phase isolation in stage I.

For cell harvesting, centrifugation is the preferred option in most cases, and if we can improve its efficiency, we may be able to reduce the cost by almost  $\sim$  54%. For cell disruption, acid hydrolysis is the preferred technology for insoluble products, and by improving the release parameters the cost can be reduced by  $\sim$ 13%. This proves that the process is more sensitive to the technology selected for cell harvesting as compared to disruption and indicates that further research and development is required for performance enhancement of harvesting technologies. For product isolation, diferential digestion of NPCM is more favorable than product solubilization. Thus, there is a need to find digestion agents which do not afect the product quality and are required in reasonable amounts.

The second study (IN SOL LQD VOL SPC) is a representative case for intracellular soluble chemicals. Separation again was the major cost contributor  $(\sim 85\%)$ . We found that biomass titer affected the optimal separation network design and overall process cost at constant product content, whereas change in product content at low biomass titers ( $\leq$ 5 g/L) did not alter the optimal design but afected the overall process cost. Further analysis suggested that simultaneous improvements in both parameters can reduce the overall process cost signifcantly.

Stage II is crucial for soluble products and the cost contribution in the overall process can vary signifcantly depending upon the available technologies. We observed that stage II could become dominant if extraction is the only feasible option. The analysis for alternate options in stage II revealed that distillation (option #1: base case) cost can increase slightly with the increase in heat of vaporization of the product (more volatile or light key component). For ATPE (option #2), it was observed that with the increase in product partition coefficient the overall cost could be decreased. For extraction (option #3), it was found that the process cost reduced signifcantly when a solvent was almost insoluble in water and was available at low cost, but the change in partition coefficient had no significant effect.

In the current studies, we include most common technologies to generate reliable process insights (e.g., identifying biomass titer, product content, separation efficiency, etc. as the key cost influencers). However, new technologies and products can be taken into account by changing model parameters and/or adding case-specifc constraints in the model. Thus, we believe that the proposed analysis will prove valuable in selecting new target bio-based chemicals that can be produced economically using microbial bioconversions and for designing costefective separation processes.

Furthermore, this study addresses the change in 'optimal' system performance. The predictions associated with the changing inputs and model parameters can assist in the development of policies or general guidelines for similar systems. Additionally, the model can be extended to account for environmental constraints with regard to waste streams or the overall environmental footprint. This can be achieved by  $(1)$  adding constraints representing environmental regulations [[147–](#page-22-24)[150](#page-22-25)] and/or (2) modifying the objective function to include an environmental sustainability indicator [[146](#page-22-26)] and employing a multi-objective optimization approach. Most importantly, it helps in identifying some promising research directions in the area of separations, a topic that has not received its due attention despite its substantial impact on the economics of biomass-to-fuels/chemicals strategies.

## **Product properties**

(In the order of classifcation)–NSL: insoluble; SOL: soluble; SLD: solid; LQD: liquid; HV: heavy; LT: light; VOL: volatile (more volatile that water); NVL: non-volatile (less volatile than water); CMD: commodity; SPC: specialty.

## **Technologies**

Ads: Adsorption; Ahy: acid hydrolysis; Atpe: aqueous two-phase extraction; Blc: bleaching; Bml: bead mill; Chr: chromatography; Chy: chemical lysis; Cnt: centrifugation; Cog: coagulation; Crs: crystallization; Ddg: diferential digestion; Dry: drying; Dst: distillation; Ely: enzyme lysis; Evp: evaporation; Ext: extraction; Flc: focculation; Flt: fotation; Ftt: fltration; Hph: homogenization; MF: microfltration; Mxs: mixer; NF: nanofltration; Prc: precipitation; Pvp: pervaporation; RO: reverse osmosis; Sdm: sedimentation; Slb: solubilization; Splt: splitter; UF: ultrafltration.

#### **Others**

CDW: cell dry weight; Kp: product partition coefficient in solvent phase for Ext technology; KpT: product partition coefficient in the top phase for Atpe technology; MINLP: mixed-integer non-linear programming; NPCM: non-product cellular materials.

#### **Additional fle**

<span id="page-19-14"></span>**[Additional fle 1.](http://dx.doi.org/10.1186/s13068-017-0804-2) A**. Information about intracellular chemicals: includes previous studies and economic assessment for some intracellular chemicals. **B**. Case study 1 – Intracellular insoluble product: includes **B.1.** Separation scheme and superstructure for intracellular insoluble product, **B.2.** Model details and equations, **B.3** Model parameters and input data, **B.4.** Integer-cuts for alternate confgurations and **B.5.** Some additional results. **C**. Case study 2 – Intracellular soluble product: includes **C.1.** Separation scheme and superstructure for intracellular soluble product, **C.2.** Model details and equations, **C.3.** Model parameters and input data, **C.4.** Integer-cuts for alternate confgurations and **C.5.** Some additional results.

#### **Author's contribution**

KMY completed the major portion of this work including the development of the case studies, mathematical modeling, optimization, analysis, and writing of the manuscript. WW helped in providing the technology models, superstructure formulation, and revision of the manuscript. CTM supervised the work and writing, and provided research direction, resources, and support for performing the study. All authors read and approved the fnal manuscript.

#### **Author details**

<sup>1</sup> Department of Chemical and Biological Engineering, University of Wisconsin-Madison, 1415 Engineering Drive, Madison, WI 53706-1691, USA. <sup>2</sup> DOE Great Lakes Bioenergy Research Center, University of Wisconsin-Madison, 1552 University Ave, Madison, WI 53726, USA.

#### **Acknowledgements**

Not applicable.

#### **Competing interests**

The authors declare that they have no competing interests.

#### **Availability of data and materials**

The additional materials such as model equations, optimization problem formulation, and data for input conditions and technology parameters are included in Additional fle [1.](#page-19-14) Also, additional results obtained from the case studies are included in Additional fle [1](#page-19-14).

#### **Funding**

This work was funded by the National Science Foundation through the Emerging Frontiers in Research and Innovation program (EFRI-1240268)

#### **Publisher's Note**

Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional afliations.

Received: 20 January 2017 Accepted: 25 April 2017 Published online: 08 May 2017

#### <span id="page-19-0"></span>**References**

- 1. Conti J, Holtberg P, Diefenderfer J, LaRose A, Turnure JT, Westfall L. International Energy Outlook 2016 with projections to 2040—US Energy Information Administration. 2016. <http://www.eia.gov/forecasts/ieo/>. Accessed 11 Jul 2016.
- 2. OECD. The Bioeconomy to 2030. Paris: Organisation for Economic Co-operation and Development; 2009. [http://www.oecd-ilibrary.org/](http://www.oecd-ilibrary.org/content/book/9789264056886-en) [content/book/9789264056886-en](http://www.oecd-ilibrary.org/content/book/9789264056886-en). Accessed 20 Jun 2016.
- <span id="page-19-1"></span>Sanders J, Scott E, Weusthuis R, Mooibroek H. Bio-refinery as the bioinspired process to bulk chemicals. Macromol Biosci. 2007;7:105–17.
- <span id="page-19-2"></span>4. Karlsson H, Ahlgren S, Sandgren M, Passoth V, Wallberg O, Hansson P-A. A systems analysis of biodiesel production from wheat straw using oleaginous yeast: process design, mass and energy balances. Biotechnol Biofuels. 2016;9:229.
- <span id="page-19-3"></span>5. Kim J, Realf MJ, Lee JH, Whittaker C, Furtner L. Design of biomass processing network for biofuel production using an MILP model. Biomass Bioenergy. 2011;35:853–71.
- <span id="page-19-4"></span>6. Davidson BH, Lievense JC. Commercializing industrial biotechnology: technology challenges and opportunities. Chem Eng Prog. 2016;6:35–42.
- <span id="page-19-5"></span>7. Marquardt W, Harwardt A, Hechinger M, Kraemer K, Viell J, Voll A. The biorenewables opportunity—toward next generation process and product systems. AIChE J. 2010;56:2228–35.
- <span id="page-19-6"></span>8. von der Assen N, Bardow A. Life cycle assessment of polyols for polyurethane production using CO<sub>2</sub> as feedstock: insights from an industrial case study. Green Chem. 2014;16:3272–80.
- 9. Jones C, Armstrong K, Ingram I, Bay E, Dodson J, Abrantes P, et al. Wider impacts: general discussion. Faraday Discuss. 2015;183:349–68.
- <span id="page-19-7"></span>10. Bender M. An overview of industrial processes for the production of olefns—C4 hydrocarbons. ChemBioEng Rev. 2014;1:136–47.
- 11. Lallemand M, Finiels A, Fajula F, Hulea V. Catalytic oligomerization of ethylene over Ni-containing dealuminated Y zeolites. Appl Catal Gen. 2006;301:196–201.
- <span id="page-19-8"></span>12. Speiser F, Braunstein P, Saussine L. Catalytic ethylene dimerization and oligomerization: recent developments with nickel complexes containing P,N-chelating ligands. Acc Chem Res. 2005;38:784–93.
- <span id="page-19-9"></span>13. Wang G, Xu C, Gao J. Study of cracking FCC naphtha in a secondary riser of the FCC unit for maximum propylene production. Fuel Process Technol. 2008;89:864–73.
- <span id="page-19-10"></span>14. Wei Y, Liu Z, Wang G, Qi Y, Xu L, Xie P, et al. Production of light olefns and aromatic hydrocarbons through catalytic cracking of naphtha at lowered temperature. In: J. Čejka NŽ and PN, editor. Studies in surface science and catalysis. Amsterdam: Elsevier; 2005. p. 1223–30. [http://](http://www.sciencedirect.com/science/article/pii/S0167299105804689) [www.sciencedirect.com/science/article/pii/S0167299105804689](http://www.sciencedirect.com/science/article/pii/S0167299105804689). Accessed 9 Dec 2016.
- <span id="page-19-11"></span>15. Brown RC, Brown TR. Biorenewable resources: engineering new products from agriculture. 2nd ed. Wiley-Blackwell; 2013. [http://www.wiley.com/](http://www.wiley.com/WileyCDA/WileyTitle/productCd-1118524950.html) [WileyCDA/WileyTitle/productCd-1118524950.html.](http://www.wiley.com/WileyCDA/WileyTitle/productCd-1118524950.html) Accessed 9 Dec 2016.
- <span id="page-19-12"></span>16. Aldor IS, Keasling JD. Process design for microbial plastic factories: metabolic engineering of polyhydroxyalkanoates. Curr Opin Biotechnol. 2003;14:475–83.
- <span id="page-19-13"></span>17. Liao JC, Mi L, Pontrelli S, Luo S. Fuelling the future: microbial engineering for the production of sustainable biofuels. Nat Rev Microbiol. 2016;14:288–304.
- 18. Lively RP, Sharma P, McCool BA, Beaudry-Losique J, Luo D, Thomas VM, et al. Anthropogenic CO2 as a feedstock for the production of algalbased biofuels. Biofuels Bioprod Biorefning. 2015;9:72–81.
- 19. Lynd LR, Weimer PJ, van Zyl WH, Pretorius IS. Microbial cellulose utilization: fundamentals and biotechnology. Microbiol Mol Biol Rev. 2002;66:506–77.
- <span id="page-20-0"></span>20. Zeng A-P, Biebl H. Bulk chemicals from biotechnology: the case of 1,3-propanediol production and the new trends. In: Scheper T, editor. Tools and applications of biochemical engineering science. Berlin Heidelberg: Springer; 2002. p. 239–59. [http://link.springer.com/chap](http://link.springer.com/chapter/10.1007/3-540-45736-4_11)[ter/10.1007/3-540-45736-4\\_11](http://link.springer.com/chapter/10.1007/3-540-45736-4_11). Accessed 1 Jun 2016.
- <span id="page-20-1"></span>21. Keasling JD. Manufacturing molecules through metabolic engineering. Science. 2010;330:1355–8.
- <span id="page-20-2"></span>22. Angermayr SA, Gorchs Rovira A, Hellingwerf KJ. Metabolic engineering of cyanobacteria for the synthesis of commodity products. Trends Biotechnol. 2015;33:352–61.
- <span id="page-20-9"></span>23. Lee JW, Na D, Park JM, Lee J, Choi S, Lee SY. Systems metabolic engineering of microorganisms for natural and non-natural chemicals. Nat Chem Biol. 2012;8:536–46.
- 24. Miskovic L, Hatzimanikatis V. Production of biofuels and biochemicals: in need of an ORACLE. Trends Biotechnol. 2010;28:391–7.
- 25. Nielsen J, Fussenegger M, Keasling J, Lee SY, Liao JC, Prather K, et al. Engineering synergy in biotechnology. Nat Chem Biol. 2014;10:319–22.
- 26. Palsson B. Metabolic systems biology. FEBS Lett. 2009;583:3900–4. 27. Quintana N, Van der Kooy F, Van de Rhee MD, Voshol GP, Verpoorte R. Renewable energy from Cyanobacteria: energy production optimiza-
- tion by metabolic pathway engineering. Appl Microbiol Biotechnol. 2011;91:471–90. 28. Stephanopoulos G, Aristidou AA, Nielsen J. Metabolic engineering:
- <span id="page-20-3"></span>principles and methodologies. Cambridge: Academic Press; 1998. 29. Knoop H, Steuer R. A computational analysis of stoichiometric con-
- straints and trade-ofs in cyanobacterial biofuel production. Synth Biol. 2015;3:47.
- <span id="page-20-4"></span>30. Salehizadeh H, Van Loosdrecht MCM. Production of polyhydroxyalkanoates by mixed culture: recent trends and biotechnological importance. Biotechnol Adv. 2004;22:261–79.
- <span id="page-20-5"></span>31. Rehm B. Microbial production of biopolymers and polymer precursors: applications and perspectives. Poole: Horizon Scientifc Press; 2009.
- <span id="page-20-6"></span>32. Fresewinkel M, Rosello R, Wilhelm C, Kruse O, Hankamer B, Posten C. Integration in microalgal bioprocess development: design of efficient, sustainable, and economic processes. Eng Life Sci. 2014;14:560–73.
- <span id="page-20-7"></span>33. Alper H, Jin Y-S, Moxley JF, Stephanopoulos G. Identifying gene targets for the metabolic engineering of lycopene biosynthesis in *Escherichia coli*. Metab Eng. 2005;7:155–64.
- 34. Tyo KE, Alper HS, Stephanopoulos GN. Expanding the metabolic engineering toolbox: more options to engineer cells. Trends Biotechnol. 2007;25:132–7.
- 35. Upadhyaya BP, DeVeaux LC, Christopher LP. Metabolic engineering as a tool for enhanced lactic acid production. Trends Biotechnol. 2014;32:637–44.
- 36. Wang L, Hatzimanikatis V. Metabolic engineering under uncertainty—II: analysis of yeast metabolism. Metab Eng. 2006;8:142–59.
- 37. Woolston BM, Edgar S, Stephanopoulos G. Metabolic Engineering: past and Future. Annu Rev Chem Biomol Eng. 2013;4:259–88.
- <span id="page-20-8"></span>38. Yadav VG, De Mey M, Giaw Lim C, Kumaran Ajikumar P, Stephanopoulos G. The future of metabolic engineering and synthetic biology: towards a systematic practice. Metab Eng. 2012;14:233–41.
- <span id="page-20-10"></span>39. Sagmeister P, Wechselberger P, Jazini M, Meitz A, Langemann T, Herwig C. Soft sensor assisted dynamic bioprocess control: efficient tools for bioprocess development. Chem Eng Sci. 2013;96:190–8.
- 40. Herwig C, Marison I, von Stockar U. On-line stoichiometry and identifcation of metabolic state under dynamic process conditions. Biotechnol Bioeng. 2001;75:345–54.
- <span id="page-20-11"></span>41. Herwig C, von Stockar U. A small metabolic fux model to identify transient metabolic regulations in Saccharomyces cerevisiae. Bioprocess Biosyst Eng. 2002;24:395–403.
- <span id="page-20-12"></span>42. Yoisalai. Biopol. Bioplastic Product and Biodegradable. 2008. [http://](http://www.bioplastic-product.com/biopol/) [www.bioplastic-product.com/biopol/](http://www.bioplastic-product.com/biopol/). Accessed 27 Jul 2016.
- <span id="page-20-13"></span>43. Mauney L. Bioplastic inventor noda presents the science behind Nodax. MHG Bio. The Biopolymer Company. 2015. [http://www.mhgbio.com/](http://www.mhgbio.com/mhgs-pha-bioplastic-inventor-dr-isao-noda-presents-science-behind-nodaxtm/) [mhgs-pha-bioplastic-inventor-dr-isao-noda-presents-science-behind](http://www.mhgbio.com/mhgs-pha-bioplastic-inventor-dr-isao-noda-presents-science-behind-nodaxtm/)[nodaxtm/](http://www.mhgbio.com/mhgs-pha-bioplastic-inventor-dr-isao-noda-presents-science-behind-nodaxtm/). Accessed 18 Aug 2016.
- <span id="page-20-14"></span>44. JFS. RIKEN and Kaneka successfully produce bioplastics from non-food Biomass: JFS Japan for Sustainability. JFS Japan for Sustainability. 2014.

[http://www.japanfs.org/en/news/archives/news\\_id034960.html.](http://www.japanfs.org/en/news/archives/news_id034960.html) Accessed 18 Aug 2016.

- <span id="page-20-15"></span>45. Brennan L, Owende P. Biofuels from microalgae—a review of technologies for production, processing, and extractions of biofuels and co-products. Renew Sustain Energy Rev. 2010;14:557–77.
- <span id="page-20-16"></span>46. Brandt K, Schembecker G. Production rate-dependent key performance indicators for a systematic design of biochemical downstream processes. Chem Eng Technol. 2016;39:354–64.
- 47. Kiss AA, Grievink J, Rito-Palomares M. A systems engineering perspective on process integration in industrial biotechnology. J Chem Technol Biotechnol. 2015;90:349–55.
- <span id="page-20-31"></span>48. Linninger AA. Industry-wide energy saving by complex separation networks. Comput Chem Eng. 2009;33:2018–27.
- <span id="page-20-17"></span>49. Shaeiwitz JA, Henry JD, Ghosh R. Bioseparation. In: Ullmann's encyclopedia of industrial chemistry. Wiley-VCH Verlag GmbH & Co.; 2000. [http://onlinelibrary.wiley.com/doi/10.1002/14356007.b03\\_11.pub2/](http://onlinelibrary.wiley.com/doi/10.1002/14356007.b03_11.pub2/abstract) [abstract](http://onlinelibrary.wiley.com/doi/10.1002/14356007.b03_11.pub2/abstract). Accessed 21 Jun 2016.
- <span id="page-20-18"></span>50. Puligundla P, Smogrovicova D, Obulam VSR, Ko S. Very high gravity (VHG) ethanolic brewing and fermentation: a research update. J Ind Microbiol Biotechnol. 2011;38:1133–44.
- <span id="page-20-19"></span>51. Xiros C, Janssen M, Byström R, Børresen BT, Cannella D, Jørgensen H, et al. Toward a sustainable biorefnery using high-gravity technology. Biofuels Bioprod Biorefning. 2017;11:15–27.
- <span id="page-20-20"></span>52. Choi J, Lee SY. Factors afecting the economics of polyhydroxyalkanoate production by bacterial fermentation. Appl Microbiol Biotechnol. 1999;51:13–21.
- <span id="page-20-21"></span>53. Choi J, Lee SY. Process analysis and economic evaluation for Poly(3 hydroxybutyrate) production by fermentation. Bioprocess Eng. 1997;17:335–42.
- <span id="page-20-22"></span>54. Frey KM, Oppermann-Sanio FB, Schmidt H, Steinbüchel A. Technicalscale production of cyanophycin with recombinant strains of *Escherichia coli*. Appl Environ Microbiol. 2002;68:3377–84.
- <span id="page-20-23"></span>55. Mooibroek H, Oosterhuis N, Giuseppin M, Toonen M, Franssen H, Scott E, et al. Assessment of technological options and economical feasibility for cyanophycin biopolymer and high-value amino acid production. Appl Microbiol Biotechnol. 2007;77:257–67.
- <span id="page-20-24"></span>56. Cuellar-Bermudez SP, Aguilar-Hernandez I, Cardenas-Chavez DL, Ornelas-Soto N, Romero-Ogawa MA, Parra-Saldivar R. Extraction and purifcation of high-value metabolites from microalgae: essential lipids, astaxanthin and phycobiliproteins. Microb Biotechnol. 2015;8:190–209.
- 57. Machmudah S, Shotipruk A, Goto M, Sasaki M, Hirose T. Extraction of Astaxanthin from *Haematococcus pluvialis* using supercritical CO<sub>2</sub> and ethanol as entrainer. Ind Eng Chem Res. 2006;45:3652–7.
- <span id="page-20-25"></span>58. Sarada R, Vidhyavathi R, Usha D, Ravishankar GA. An efficient method for extraction of astaxanthin from green alga *Haematococcus pluvialis*. J Agric Food Chem. 2006;54:7585–8.
- <span id="page-20-26"></span>59. Ruiz-Ruiz F, Benavides J, Rito-Palomares M. Scaling-up of a B-phycoerythrin production and purifcation bioprocess involving aqueous two-phase systems: practical experiences. Process Biochem. 2013;48:738–45.
- <span id="page-20-27"></span>60. Clarkson AI, Bogle IDL, Titchener-Hooker NJ. Modelling and simulation of fractional precipitation—comparison with pilot plant data. Comput Chem Eng. 1992;16(Supplement 1):S441–7.
- 61. Polykarpou EM, Dalby PA, Papageorgiou LG. Optimal synthesis of chromatographic trains for downstream protein processing. Biotechnol Prog. 2011;27:1653–60.
- 62. Schenk M, Gani R, Bogle D, Pistikopoulos EN. A hybrid modelling approach for separation systems involving distillation. Chem Eng Res Des. 1999;77:519–34.
- <span id="page-20-28"></span>63. Winkelnkemper T, Schembecker G. Purifcation performance index and separation cost indicator for experimentally based systematic downstream process development. Sep Purif Technol. 2010;72:34–9.
- <span id="page-20-29"></span>64. Molina Grima E, Belarbi E-H, Acién Fernández FG, Robles Medina A, Chisti Y. Recovery of microalgal biomass and metabolites: process options and economics. Biotechnol Adv. 2003;20:491–515.
- <span id="page-20-30"></span>65. Chhatre S, Francis R, Newcombe AR, Zhou Y, Titchener-Hooker N, King J, et al. Global sensitivity analysis for the determination of parameter importance in bio-manufacturing processes. Biotechnol Appl Biochem. 2008;51:79–90.
- 66. Hamby DM. A review of techniques for parameter sensitivity analysis of environmental models. Environ Monit Assess. 1994;32:135–54.
- <span id="page-21-0"></span>67. Zhang Y, Dubé MA, McLean DD, Kates M. Biodiesel production from waste cooking oil: 2. Economic assessment and sensitivity analysis. Bioresour Technol. 2003;90:229–40.
- <span id="page-21-1"></span>68. Lakhdar K, Zhou Y, Savery J, Titchener-Hooker NJ, Papageorgiou LG. Medium term planning of biopharmaceutical manufacture using mathematical programming. Biotechnol Prog. 2005;21:1478–89.
- 69. Natali JM, Pinto JM, Papageorgiou LG. Efficient MILP formulations for the simultaneous optimal peptide tag design and downstream processing synthesis. AIChE J. 2009;55:2303–17.
- 70. Polykarpou EM, Dalby PA, Papageorgiou LG. An MILP formulation for the synthesis of protein purifcation processes. Chem Eng Res Des. 2012;90:1262–70.
- 71. Ruiz GJ, Kim SB, Moon J, Zhang L, Linninger AA. Design and optimization of energy efficient complex separation networks. Comput Chem Eng. 2010;34:1556–63.
- <span id="page-21-7"></span>72. Simeonidis E, Pinto JM, Papageorgiou LG. An MILP model for optimal design of purifcation tags and synthesis of downstream processing. In: Espuña LP and A, editor. Computer Aided Chemical Engineering. Amsterdam: Elsevier; 2005. p. 1537–42. [http://www.sciencedirect.com/](http://www.sciencedirect.com/science/article/pii/S1570794605800987) [science/article/pii/S1570794605800987](http://www.sciencedirect.com/science/article/pii/S1570794605800987). Accessed 16 Jan 2017.
- 73. Zhang L, Linninger AA. Towards computer-aided separation synthesis. AIChE J. 2006;52:1392–409.
- <span id="page-21-4"></span>74. Kokossis AC, Tsakalova M, Pyrgakis K. Design of integrated biorefneries. Comput Chem Eng. 2015;81:40–56.
- 75. Alva-Argáez A, Kokossis AC, Smith R. Wastewater minimisation of industrial systems using an integrated approach. Comput Chem Eng. 1998;22:S741–4.
- <span id="page-21-2"></span>76. Alva-Argáez A, Vallianatos A, Kokossis A. A multi-contaminant transhipment model for mass exchange networks and wastewater minimisation problems. Comput Chem Eng. 1999;23:1439–53.
- <span id="page-21-3"></span>77. Pavia DL. Introduction to organic laboratory techniques: a small scale approach. Boston: Cengage Learning; 2005.
- <span id="page-21-5"></span>78. Yeomans H, Grossmann IE. A systematic modeling framework of superstructure optimization in process synthesis. Comput Chem Eng. 1999;23:709–31.
- <span id="page-21-8"></span>79. Rizwan M, Lee JH, Gani R. Optimal processing pathway for the production of biodiesel from microalgal biomass: a superstructure based approach. Comput Chem Eng. 2013;58:305–14.
- <span id="page-21-37"></span>80. Wai PPC, Bogle IDL, Bagherpour K, Gani R. Process synthesis and simulation strategies for integrated biochemical process design. Comput Chem Eng. 1996;20:S357–62.
- <span id="page-21-6"></span>81. Wu W, Henao CA, Maravelias CT. A superstructure representation, generation, and modeling framework for chemical process synthesis. AIChE J. 2016.<http://onlinelibrary.wiley.com/doi/10.1002/aic.15300/abstract>. Accessed 3 Aug 2016.
- <span id="page-21-9"></span>82. Belter PA, Cussler EL, Hu W-S. Bioseparations: downstream processing for biotechnology. USA: Wiley; 1988.
- 83. Goldberg E. Handbook of downstream processing. London: Blackie Academic & Professional; 1997.
- <span id="page-21-36"></span>84. Harrison RG, Todd P, Rudge SR, Petrides DP. Bioseparations science and engineering. Oxford: Oxford University; 2003.
- <span id="page-21-30"></span>85. Kennedy JF, Cabral JMS, editors. Recovery processes for biological materials. 1st ed. Chichester; New York: Wiley; 1993.
- <span id="page-21-31"></span>86. Verrall M. Downstream processing of natural products: a practical handbook. United Kingdom: Wiley; 1996. [http://www.wiley.com/WileyCDA/](http://www.wiley.com/WileyCDA/WileyTitle/productCd-0471963267.html) [WileyTitle/productCd-0471963267.html](http://www.wiley.com/WileyCDA/WileyTitle/productCd-0471963267.html). Accessed 2 Oct 2015.
- 87. Sikdar SK, Burckle J, Rogut J. Separation methods for environmental technologies. Environ Prog. 2001;20:1–11.
- 88. Sikdar SK, Todd PW, Bier M. Frontiers in Bioprocessing. Boca Raton: CRC Press; 1989.
- 89. Sikdar SK. Sustainability perspective and chemistry-based technologies. Ind Eng Chem Res. 2007;46:4727–33.
- 90. Asenjo JA. Separation processes in biotechnology. New York: Marcel Dekker Inc.; 1990.
- 91. Asenjo JA, Andrews BA. Challenges and trends in bioseparations. J Chem Technol Biotechnol. 2008;83:117–20.
- 92. Lienqueo ME, Asenjo JA. Use of expert systems for the synthesis of downstream protein processes. Comput Chem Eng. 2000;24:2339–50.
- 93. Zydney AL. Continuous downstream processing for high value biological products: a review. Biotechnol Bioeng. 2016;113:465–75.
- 94. Van Reis R, Zydney A. Bioprocess membrane technology. J Membr Sci. 2007;297:16–50.
- <span id="page-21-10"></span>95. Zydney AL. Membrane technology for purifcation of therapeutic proteins. Biotechnol Bioeng. 2009;103:227–30.
- <span id="page-21-11"></span>96. Yenkie KM, Wu W, Clark RL, Pfeger BF, Root TW, Maravelias CT. A roadmap for the synthesis of separation networks for the recovery of bio-based chemicals: matching biological and process feasibility. Biotechnol Adv. 2016. doi[:10.1016/j.biotechadv.2016.10.003](http://dx.doi.org/10.1016/j.biotechadv.2016.10.003).
- <span id="page-21-12"></span>97. Wu W, Yenkie KM, Maravelias CT. A superstructure-based framework for bio-separation network synthesis. Comput Chem Eng. 2017;96:1–17.
- <span id="page-21-13"></span>Ulrich GD, Vasudevan PT. Chemical engineering process design and economics: a practical guide. Yerevan: Process Pub; 2004.
- <span id="page-21-14"></span>99. Tawarmalani M, Sahinidis NV. A polyhedral branch-and-cut approach to global optimization. Math Program. 2005;103:225–49.
- <span id="page-21-15"></span>100. Petrides DP. Intelligen, Inc.: SuperPro designer, batch process simulation, environmental impact assessment. Intelligen Inc; 2013. [http://](http://www.intelligen.com/superpro_overview.html) [www.intelligen.com/superpro\\_overview.html](http://www.intelligen.com/superpro_overview.html). Accessed 5 May 2016.
- <span id="page-21-16"></span>101. Anis SNS, Iqbal NM, Kumar S, Al-Ashraf A. Increased recovery and improved purity of PHA from recombinant Cupriavidus necator. Bioengineered. 2013;4:115–8.
- <span id="page-21-21"></span>102. Lodish H, Berk A, Zipursky SL, Matsudaira P, Baltimore D, Darnell J. Molecular cell biology. 4th ed. New York: W. H. Freeman; 2000.
- <span id="page-21-22"></span>Armstrong GA, Hearst JE. Carotenoids 2: genetics and molecular biology of carotenoid pigment biosynthesis. FASEB J. 1996;10:228–37.
- <span id="page-21-23"></span>104. Cooney M. Case studies of separation in biorefneries—extraction of algae oil from microalgae. In: Ramaswamy S, Huang H-J, Ramarao BV, editors. Separation and purifcation technologies in biorefneries. Hoboken: Wiley; 2013. p. 533–54. [http://onlinelibrary.wiley.com.ezproxy.library.wisc.](http://onlinelibrary.wiley.com.ezproxy.library.wisc.edu/doi/10.1002/9781118493441.ch21/summary) [edu/doi/10.1002/9781118493441.ch21/summary](http://onlinelibrary.wiley.com.ezproxy.library.wisc.edu/doi/10.1002/9781118493441.ch21/summary). Accessed 13 Oct 2015.
- <span id="page-21-24"></span>105. Rodolf L, Chini Zittelli G, Bassi N, Padovani G, Biondi N, Bonini G, et al. Microalgae for oil: strain selection, induction of lipid synthesis and outdoor mass cultivation in a low-cost photobioreactor. Biotechnol Bioeng. 2009;102:100–12.
- <span id="page-21-17"></span>106. Choi J, Lee SY. Efficient and economical recovery of poly(3-hydroxybutyrate) from recombinant Escherichia coli by simple digestion with chemicals. Biotechnol Bioeng. 1999;62:546–53.
- <span id="page-21-18"></span>107. Sayyed RZ, Gangurde NS, Chincholkar SB. Hypochlorite digestion method for efficient recovery of PHB from Alcaligenes faecalis. Indian J Microbiol. 2009;49:230–2.
- <span id="page-21-20"></span><span id="page-21-19"></span>108. Epp KA, Fetterolf DM. Recovery Processes. J Valid Technol. 2009;14:43–8. 109. Singh SM, Panda AK. Solubilization and refolding of bacterial inclusion
- body proteins. J Biosci Bioeng. 2005;99:303–10. 110. Kim J, Sen SM, Maravelias CT. An optimization-based assessment
- <span id="page-21-25"></span>framework for biomass-to-fuel conversion strategies. Energy Environ Sci. 2013;6:1093.
- <span id="page-21-26"></span>111. Patil KR, Åkesson M, Nielsen J. Use of genome-scale microbial models for metabolic engineering. Curr Opin Biotechnol. 2004;15:64–9.
- <span id="page-21-27"></span>112. Runguphan W, Keasling JD. Metabolic engineering of Saccharomyces cerevisiae for production of fatty acid-derived biofuels and chemicals. Metab Eng. 2014;21:103–13.
- 113. Spadiut O, Rittmann S, Dietzsch C, Herwig C. Dynamic process conditions in bioprocess development. Eng Life Sci. 2013;13:88–101.
- <span id="page-21-28"></span>114. Wechselberger P, Sagmeister P, Herwig C. Real-time estimation of biomass and specifc growth rate in physiologically variable recombinant fed-batch processes. Bioprocess Biosyst Eng. 2013;36:1205–18.
- <span id="page-21-29"></span>115. Fischer CR, Klein-Marcuschamer D, Stephanopoulos G. Selection and optimization of microbial hosts for biofuels production. Metab Eng. 2008;10:295–304.
- <span id="page-21-32"></span>116. Greenwell HC, Laurens LML, Shields RJ, Lovitt RW, Flynn KJ. Placing microalgae on the biofuels priority list: a review of the technological challenges. J R Soc Interface. 2009; rsif20090322.
- <span id="page-21-33"></span>117. Brennan L, Owende P. Biofuels from microalgae—a review of technologies for production, processing, and extractions of biofuels and co-products. Renew Sustain Energy Rev. 2010;14:557–77.
- <span id="page-21-34"></span>118. Aikawa S, Nishida A, Ho S-H, Chang J-S, Hasunuma T, Kondo A. Glycogen production for biofuels by the euryhaline cyanobacteria *Synechococcus* sp. strain PCC 7002 from an oceanic environment. Biotechnol Biofuels. 2014;7:88.
- <span id="page-21-35"></span>119. Castanha RF, de Morais LAS, Mariano AP, Monteiro RTR. Comparison of two lipid extraction methods produced by yeast in cheese whey. Braz Arch Biol Technol. 2013;56:629–36.
- <span id="page-22-2"></span>120. Li Q, Du W, Liu D. Perspectives of microbial oils for biodiesel production. Appl Microbiol Biotechnol. 2008;80:749–56.
- <span id="page-22-3"></span>121. Ambler CM. The fundamentals of separation, including Sharples "Sigma value" for predicting equipment performance. Ind Eng Chem Res. 1961;53:430–3.
- <span id="page-22-4"></span>122. Miranda JR, Passarinho PC, Gouveia L. Pre-treatment optimization of *Scenedesmus obliquus* microalga for bioethanol production. Bioresour Technol. 2012;104:342–8.
- <span id="page-22-5"></span>123. Chisti Y, Moo-Young M. Disruption of microbial cells for intracellular products. Enzyme Microb Technol. 1996;8:194–204.
- <span id="page-22-0"></span>124. Günerken E, D'Hondt E, Eppink MHM, Garcia-Gonzalez L, Elst K, Wijffels RH. Cell disruption for microalgae biorefneries. Biotechnol Adv. 2015;33:243–60.
- <span id="page-22-1"></span>125. Andrews BA, Asenjo JA. Enzymatic lysis and disruption of microbial cells. Trends Biotechnol. 1987;5:273–7.
- <span id="page-22-6"></span>126. Kontoravdi C, Samsatli NJ, Shah N. Development and design of biopharmaceutical processes. Curr Opin Chem Eng. 2013;2:435–41.
- <span id="page-22-11"></span>127. Schwenzfeier A, Wierenga PA, Gruppen H. Isolation and characterization of soluble protein from the green microalgae Tetraselmis sp. Bioresour Technol. 2011;102:9121–7.
- <span id="page-22-12"></span>128. Titchener-Hooker NJ, Hoare M. Recovery of intracellular products. In: Costa CA, Cabral JS, editors. Chromatographic and membrane processes in biotechnology. Netherlands: Springer; 1991. p. 253–66. [http://](http://link.springer.com/chapter/10.1007/978-94-011-3470-5_13) [link.springer.com/chapter/10.1007/978-94-011-3470-5\\_13](http://link.springer.com/chapter/10.1007/978-94-011-3470-5_13). Accessed 16 Sep 2016.
- <span id="page-22-13"></span>129. Davis R, Aden A, Pienkos PT. Techno-economic analysis of autotrophic microalgae for fuel production. Appl Energy. 2011;88:3524–31.
- <span id="page-22-14"></span>130. Koo TY, Park TH. Expression of recombinant human growth hormone in a soluble form in *Escherichia coli* by slowing down the protein synthesis rate. J Microbiol Biotechnol. 2007;17:579–85.
- <span id="page-22-7"></span>131. Agbor VB, Cicek N, Sparling R, Berlin A, Levin DB. Biomass pretreatment: fundamentals toward application. Biotechnol Adv. 2011;29:675–85.
- <span id="page-22-8"></span>132. Asenjo JA, Andrews BA. Aqueous two-phase systems for protein separation: phase separation and applications. J Chromatogr A. 2012;1238:1–10.
- <span id="page-22-9"></span>133. Benavides J, Rito-Palomares M, Asenjo JA. Aqueous two-phase systems. In: Moo-Young M, editor. Comprehensive biotechnology (Second Edition). Burlington: Academic Press; 2011. p. 697–713. [http://www.sci](http://www.sciencedirect.com/science/article/pii/B9780080885049001240)[encedirect.com/science/article/pii/B9780080885049001240.](http://www.sciencedirect.com/science/article/pii/B9780080885049001240) Accessed 12 Apr 2016.
- <span id="page-22-10"></span>134. Zhang J, Hu B. Liquid–liquid extraction (LLE). In: Ramaswamy S, Huang H-J, Ramarao BV, editors. Separation and purifcation technologies in biorefneries. Hoboken: Wiley; 2013. p. 61–78. [http://onlinelibrary.wiley.](http://onlinelibrary.wiley.com.ezproxy.library.wisc.edu/doi/10.1002/9781118493441.ch3/summary) [com.ezproxy.library.wisc.edu/doi/10.1002/9781118493441.ch3/sum](http://onlinelibrary.wiley.com.ezproxy.library.wisc.edu/doi/10.1002/9781118493441.ch3/summary)[mary](http://onlinelibrary.wiley.com.ezproxy.library.wisc.edu/doi/10.1002/9781118493441.ch3/summary). Accessed 13 Oct 2015.
- <span id="page-22-15"></span>135. Towler G, Sinnott RK. Chemical engineering design, second edition: principles, practice and economics of plant and process design. 2nd ed. Boston, MA: Butterworth-Heinemann; 2012.
- <span id="page-22-16"></span>136. Riesenberg D, Guthke R. High-cell-density cultivation of microorganisms. Appl Microbiol Biotechnol. 1999;51:422–30.
- <span id="page-22-17"></span>137. Hatti-Kaul R, Mattiasson B. Isolation and Purifcation of Proteins. New York: Marcel Dekker Inc.; 2003.
- <span id="page-22-18"></span>138. Kathiresan S, Sarada R, Bhattacharya S, Ravishankar GA. Culture media optimization for growth and phycoerythrin production from Porphyridium purpureum. Biotechnol Bioeng. 2007;96:456–63.
- <span id="page-22-19"></span>139. Gorak A, Sorensen E. Distillation: fundamentals and principles. Cambridge: Academic Press; 2014.
- 140. Lei Z, Chen B. Distillation. In: Ramaswamy S, Huang H-J, Ramarao BV, editors. Separation and purifcation technologies in biorefneries. Hoboken: Wiley; 2013. p. 37–60. [http://onlinelibrary.wiley.com.ezproxy.library.](http://onlinelibrary.wiley.com.ezproxy.library.wisc.edu/doi/10.1002/9781118493441.ch2/summary) [wisc.edu/doi/10.1002/9781118493441.ch2/summary.](http://onlinelibrary.wiley.com.ezproxy.library.wisc.edu/doi/10.1002/9781118493441.ch2/summary) Accessed 13 Oct 2015.
- <span id="page-22-20"></span>141. Seader JD, Henley EJ, Roper DK. Separation process principles with applications using process simulators. 4th ed. Hoboken: Wiley; 2010.
- 142. Hatti-Kaul R. Aqueous two-phase systems. Mol Biotechnol. 2001;19:269–77.
- <span id="page-22-21"></span>143. Sikdar SK, Cole KD, Stewart RM, Szlag DC, Todd P, Cabezas H. Aqueous two-phase extraction in bioseparations: an assessment. Nat Biotechnol. 1991;9:252–6.
- <span id="page-22-22"></span>144. Koch J, Shiveler G. Design principles for liquid–liquid extraction. chemical engineering progress. 2015. [http://www.aiche.org/resources/publi](http://www.aiche.org/resources/publications/cep/2015/november/design-principles-liquid-liquid-extraction)[cations/cep/2015/november/design-principles-liquid-liquid-extraction](http://www.aiche.org/resources/publications/cep/2015/november/design-principles-liquid-liquid-extraction). Accessed 12 Apr 2016.
- <span id="page-22-23"></span>145. Robl S, Gou L, Gere A, Sordo M, Lorenz H, Mayer A, et al. Chiral separation by combining pertraction and preferential crystallization. Chem Eng Process Process Intensif. 2013;67:80–8.
- <span id="page-22-26"></span>146. Doshi R, Diwekar U, Benavides PT, Yenkie KM, Cabezas H. Maximizing sustainability of ecosystem model through socio-economic policies derived from multivariable optimal control theory. Clean Technol Environ Policy. 2015;17:1573–83.
- <span id="page-22-24"></span>147. Fischer G, Prieler S, van Velthuizen H, Berndes G, Faaij A, Londo M, et al. Biofuel production potentials in Europe: sustainable use of cultivated land and pastures, Part II: Land use scenarios. Biomass Bioenergy. 2010;34:173–87.
- 148. Beringer T, Lucht W, Schaphoff S. Bioenergy production potential of global biomass plantations under environmental and agricultural constraints. GCB Bioenergy. 2011;3:299–312.
- 149. Tilman D, Socolow R, Foley JA, Hill J, Larson E, Lynd L, et al. Benefcial biofuels—the food, energy, and environment trilemma. Science. 2009;325:270–1.
- <span id="page-22-25"></span>150. Rhodes JS, Keith DW. Biomass with capture: negative emissions within social and environmental constraints: an editorial comment. Clim Change. 2008;87:321–8.

## Submit your next manuscript to BioMed Central and we will help you at every step:

- **•** We accept pre-submission inquiries
- **•** Our selector tool helps you to find the most relevant journal
- We provide round the clock customer support
- Convenient online submission
- **•** Thorough peer review
- Inclusion in PubMed and all major indexing services
- Submit your manuscript at www.biomedcentral.com/submit

**•** Maximum visibility for your research

![](_page_22_Picture_41.jpeg)